

Problem #1: see lecture notes

Problem #2: values in textbook front cover

HW#2 Solutions

#24 $E = \frac{hc}{\lambda} = hf$ a) $\frac{1240 \text{ eV}\cdot\text{nm}}{380 \text{ nm}} = \boxed{3.26 \text{ eV}}$ $\frac{1240 \text{ eV}\cdot\text{nm}}{750 \text{ nm}} = \boxed{1.65 \text{ eV}}$

b) $\frac{1240 \text{ eV}\cdot\text{nm}}{0.30 \text{ m/ns}} \cdot 100 \text{ MHz} = \boxed{3.26 \text{ eV}}$
 $\boxed{4.13 \times 10^{-7} \text{ eV}}$ note: $\lambda = \frac{c}{f} = \frac{0.30 \text{ m/ns}}{100 \text{ MHz}} = 3.0 \text{ m}$

#26 $eV_0 = hf - \phi$ a) threshold frequency f_0 when $V_0 = 0$
 $f = \frac{\phi}{h} = \frac{1.9 \text{ eV} \cdot 0.3 \text{ m/ns}}{1240 \text{ eV}\cdot\text{nm}} = \boxed{4.59 \times 10^{14} \text{ Hz}}$
 $\lambda = \frac{c}{f} = \frac{hc}{\phi} = \frac{1240 \text{ eV}\cdot\text{nm}}{1.9 \text{ eV}} = \boxed{653 \text{ nm}}$

b) stopping potential when $\lambda = 300, 400 \text{ nm}$.
 $V_0 = (hf - \phi)/e = (\frac{hc}{\lambda} - \phi)/e = (\frac{1240 \text{ eV}\cdot\text{nm}}{300 \text{ nm}} - 1.9 \text{ eV})/e = \boxed{2.23 \text{ V}}$
 $400 \text{ nm} \rightarrow \boxed{1.20 \text{ V}}$

#31 $E = n \cdot hf = 60 \cdot \frac{1240 \text{ eV}\cdot\text{nm}}{550 \text{ nm}} = 135 \text{ eV}$
 $= \frac{n hc}{\lambda} = 135 \times 1.6 \times 10^{-19} \text{ C}\cdot\text{V} = \boxed{2.17 \times 10^{-17} \text{ J}}$

#40 calibration of Bragg spectrometer: $\theta_B = 6^\circ 42' = 6.7^\circ$

$$2d \sin \theta = m \lambda$$

$$\frac{2d}{m} = \frac{\lambda}{\sin \theta} = \frac{0.0711 \text{ nm}}{\sin(6.7^\circ)} = 0.6094 \text{ nm}$$

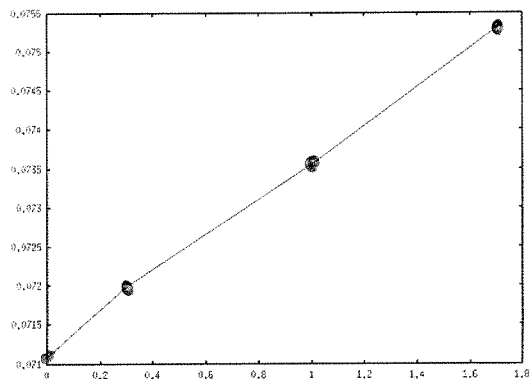
θ_c	$e_1 = 1 - \cos \theta_c$	θ_B	$\lambda_2 = (\frac{2d}{m}) \sin \theta_B$
0	0	6°42'	0.0711 nm
45°	0.293	6°47'	0.0720
90°	1	6°56'	0.0736
135°	1.707	7°06'	0.0753

y-intercept = 0.07117 nm

slope = $\boxed{0.002432 \text{ nm}}$

= λ_c

$\chi^2 = 1.59 \times 10^{-8} \text{ nm} \rightarrow \delta y = \sqrt{\frac{\chi^2}{2}} \approx 0.0009 \text{ nm}$



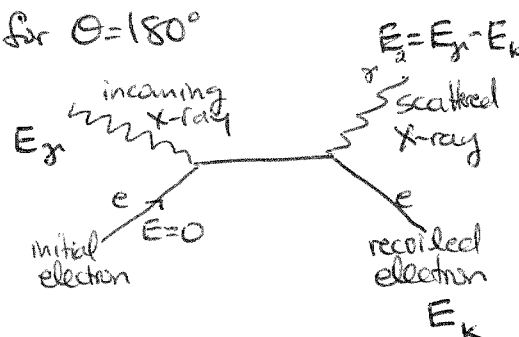
#41. a) $\lambda_c = \frac{hc}{m_e c^2} = \frac{1240 \text{ MeV} \cdot \text{fm}}{938.2 \text{ MeV}} = \boxed{1.321 \text{ fm}}$ (1 fm = 10^{-15} m)

b) $E = \frac{hc}{\lambda_c} = m_e c^2$ electron: $m_e c^2 = \boxed{0.511 \text{ MeV}}$ proton: $m_p c^2 = \boxed{938.2 \text{ MeV}}$
 both are gamma rays.

(not assigned)

#47. $eV_0 = hf = \frac{hc}{\lambda}$ $\Rightarrow V_0 = \frac{hc}{e} \cdot \frac{m}{2d \sin \theta} = \frac{1240 \text{ eV} \cdot \text{nm}}{e} \cdot \frac{1}{2 \cdot 0.28 \text{ nm} \sin(20^\circ)} = \boxed{647 \text{ kV}}$
 $2d \sin \theta = m\lambda$

#49. Compton: $\lambda_2 - \lambda_1 = \frac{h}{m_e c} (1 - \cos \theta) = \frac{h \cdot 2}{m_e c}$ for $\theta = 180^\circ$
 using $E = \frac{hc}{\lambda}$, $\frac{hc}{E_2} - \frac{hc}{E_1} = \frac{2hc}{m_e c^2}$



using conservation of energy
 $\frac{1}{E_x - E_k} - \frac{1}{E_x} = \frac{2}{m_e c^2}$

Solving for E_k :

$$\boxed{E_k = \frac{2E_x^2}{2E_x + m_e c^2}} = \frac{(hf)^2}{(hf) + \frac{1}{2}m_e c^2} = \boxed{\frac{hf}{1 + \frac{m_e c^2}{2hf}}}$$