

L58-Pauli Exclusion Principle

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* Quantum mechanical multi-particle states:

- for indistinguishable particles, $P(\vec{r}_1, \vec{r}_2) = P(\vec{r}_2, \vec{r}_1)$

define P_{12} "exchange parity" operator: $P_{12} \Psi(\vec{r}_1, \vec{r}_2) \equiv \Psi(\vec{r}_2, \vec{r}_1)$

$$\text{if } \boxed{P_{12} \Psi = \pm \Psi} \text{ then } P_{\pm}(\vec{r}_2, \vec{r}_1) = (P_{12} \Psi)^* (P_{12} \Psi) = \pm \Psi^* \cdot \pm \Psi = P_{\pm}(\vec{r}_1, \vec{r}_2)$$

so the probability is invariant under particle exchange.

Solution: let $\Psi_{\pm} = A(\Psi \pm P_{12} \Psi)$ [anti]symmetrized wave function

$$\text{then } P_{12} \Psi_{\pm} = A(P_{12} \Psi \pm P_{12} P_{12} \Psi) = \pm A(\Psi \pm P_{12} \Psi) = \pm \Psi_{\pm}$$

$$\Psi_{\pm}(\vec{r}_1, \vec{r}_2) = A(\Psi_a(\vec{r}_1) \Psi_b(\vec{r}_2) \pm \Psi_b(\vec{r}_1) \Psi_a(\vec{r}_2))$$

$$\text{normalization: } \langle \Psi_{\pm} | \Psi_{\pm} \rangle = |A|^2 \left(\langle \Psi_a | \Psi_a \rangle_1 \langle \Psi_b | \Psi_b \rangle_2 \pm \langle \Psi_a | \Psi_b \rangle_1 \langle \Psi_b | \Psi_a \rangle_2 \right. \\ \left. \pm \langle \Psi_b | \Psi_a \rangle_1 \langle \Psi_a | \Psi_b \rangle_2 + \langle \Psi_b | \Psi_b \rangle_1 \langle \Psi_a | \Psi_a \rangle_2 \right) = 2.$$

$$\langle \Psi_a | \Psi_a \rangle = 1 \quad \langle \Psi_a | \Psi_b \rangle = 0$$

- spin-statistics theorem: P_{12} (particle exchange) has the same properties as rotating the relative co-ordinate space by 360° for half-integer spin, this incurs a (-) sign

+ : Bosons particles with integer spin $S=0, 1, 2, \dots$
- : Fermions " " half-integer spin $S=\frac{1}{2}, \frac{3}{2}, \dots$

for a hand-waving argument involving a "perfect murder" with "twists in the plot", see http://math.ucr.edu/home/baez/spin_stat.html

- * Pauli Exclusion principle: no two identical fermions can occupy the same single-particle state: ie $\Psi_a = \Psi_b$

$$\Psi_{\pm}(\vec{r}_1, \vec{r}_1) = A(\Psi_a(\vec{r}_1) \Psi_a(\vec{r}_1) - \Psi_a(\vec{r}_1) \Psi_a(\vec{r}_1)) = 0$$

* note that $[H, P_{12}] = 0$ since $H_{12} = H_{21}$ and $H_{12}P_{12} = P_{12}H_{21}$
thus particle exchange parity is a constant of the motion

[anti]symmetric states are always [anti]symmetric.

- * Slater determinant - used to antisymmetrize an n-particle wavefunction built from single-particle states:
- given the 3 (or n) single-particle wave-functions $\psi_a(x), \psi_b(x), \psi_c(x)$

$$\text{let } \Psi_{abc}(x_1, x_2, x_3) = \frac{1}{\sqrt{3!}} \begin{vmatrix} \psi_a(x_1) & \psi_a(x_2) & \psi_a(x_3) \\ \psi_b(x_1) & \psi_b(x_2) & \psi_b(x_3) \\ \psi_c(x_1) & \psi_c(x_2) & \psi_c(x_3) \end{vmatrix} \quad [\text{determinant has 6 terms}]$$

$$= \frac{1}{\sqrt{3}} \left(\begin{matrix} \psi_a(x_1) \psi_b(x_2) \psi_c(x_3) \\ -\psi_a(x_1) \psi_b(x_3) \psi_c(x_2) \end{matrix} + \dots + a \rightarrow b \rightarrow c \rightarrow a - a \rightarrow c \rightarrow b \rightarrow a - \dots \right)$$

then $P_{23} \Psi_{abc} = \begin{bmatrix} \text{switch top and} \\ \text{bottom rows} \end{bmatrix} = -\Psi_{abc}$

likewise, $P_{12} \Psi_{abc} = P_{23} \Psi_{abc} = P_{31} \Psi_{abc} = -\Psi_{abc}$

i.e. Ψ_{abc} is completely antisymmetric, as required by the Pauli Exclusion Principle.

- * Exchange "Force": calculate $\langle (x_1 - x_2)^2 \rangle = \langle x_1^2 \rangle + \langle x_2^2 \rangle - 2 \langle x_1 x_2 \rangle$

$$\text{for } \Psi(x_1, x_2) = \psi_a(x_1) \psi_b(x_2) \Rightarrow \Psi_{\pm}(x_1, x_2) = \frac{1}{\sqrt{2}} (\Psi(x_1, x_2) \pm \Psi(x_2, x_1))$$

A) distinguishable particles:

$$\langle x_1^2 \rangle = \int d\mathbf{x}_1 d\mathbf{x}_2 \Psi_a^*(x_1) \Psi_b^*(x_2) x_1^2 \Psi_a(x_1) \Psi_b(x_2) = \int d\mathbf{x}_1 \Psi_a^* x_1^2 \Psi_a \cdot \int d\mathbf{x}_2 \Psi_b^* \Psi_b = \langle x^2 \rangle_a$$

$$\langle x_2^2 \rangle = \dots = \int d\mathbf{x}_1 \Psi_a^* \Psi_a \cdot \int d\mathbf{x}_2 \Psi_b^* x_2^2 \Psi_b = \langle x^2 \rangle_b$$

$$\langle x_1 x_2 \rangle = \dots = \int d\mathbf{x}_1 \Psi_a^* x_1 \Psi_a \cdot \int d\mathbf{x}_2 \Psi_b^* x_2 \Psi_b = \langle x \rangle_a \langle x \rangle_b$$

$$\boxed{\langle (x_1 - x_2)^2 \rangle = \langle x^2 \rangle_a - 2 \langle x \rangle_a \langle x \rangle_b + \langle x^2 \rangle_b}$$

B) identical particles:

$$\begin{aligned}
 \langle x_1^2 \rangle &= \int d\mathbf{x}_1 d\mathbf{x}_2 \frac{1}{\sqrt{2}} (\Psi_{ab}^*(x_1, x_2) \pm \Psi_{ab}^*(x_2, x_1)) x_1^2 \frac{1}{\sqrt{2}} (\Psi_{ab}(x_1, x_2) \pm \Psi_{ab}(x_2, x_1)) \\
 &= \frac{1}{2} \left[\int d\mathbf{x}_1 \Psi_a^* x_1^2 \Psi_a \cdot \int d\mathbf{x}_2 \Psi_b^* \Psi_b + \int d\mathbf{x}_1 \Psi_b^* x_1^2 \Psi_b \cdot \int d\mathbf{x}_2 \Psi_a^* \Psi_a \right] \\
 &\quad \pm \frac{1}{2} \left[\int d\mathbf{x}_1 \Psi_a^* x_1^2 \Psi_b \cdot \int d\mathbf{x}_2 \Psi_b^* \Psi_a \stackrel{\text{go}}{=} \text{+ complex conjugate} \right] \\
 &= \frac{1}{2} [\langle x_1^2 \rangle_a + \langle x_1^2 \rangle_b] \quad \text{likewise } \langle x_2^2 \rangle = \frac{1}{2} [\langle x_2^2 \rangle_a + \langle x_2^2 \rangle_b] = \langle x_1^2 \rangle
 \end{aligned}$$

$$\begin{aligned}
 \langle xx_2 \rangle &= \frac{1}{2} \left[\int d\mathbf{x}_1 \Psi_a^* x_1 \Psi_a \cdot \int d\mathbf{x}_2 \Psi_b^* x_2 \Psi_b + \int d\mathbf{x}_1 \Psi_b^* x_1 \Psi_b \cdot \int d\mathbf{x}_2 \Psi_a^* x_2 \Psi_a \right] \\
 &\quad \pm \frac{1}{2} \left[\underbrace{\int d\mathbf{x}_1 \Psi_a^* x_1 \Psi_b}_{\langle x \rangle_{ab}} \cdot \underbrace{\int d\mathbf{x}_2 \Psi_b^* x_2 \Psi_a^*}_{\langle x \rangle_{ab}^*} + \int d\mathbf{x}_1 \Psi_b^* x_1 \Psi_a \cdot \int d\mathbf{x}_2 \Psi_a^* x_2 \Psi_b \right] \\
 &= \langle x \rangle_a \langle x_a \rangle \pm |\langle x \rangle_{ab}|^2
 \end{aligned}$$

thus $\langle (x_1 - x_2)^2 \rangle = \underbrace{\langle x_1^2 \rangle_a - 2 \langle x \rangle_a \langle x \rangle_b + \langle x_2^2 \rangle_b}_{\langle (x_1 - x_2)^2 \rangle \text{ distinguishable particles.}} \mp 2 \underbrace{|\langle x \rangle_{ab}|^2}_{\text{bosons/fermions.}}$

* example: H_2 covalent bond

$$\Psi(\vec{r}_1, \vec{r}_2) = \underbrace{\Psi(\vec{R})}_{\text{CM, free}} \cdot \underbrace{\Psi(\vec{r})}_{\text{relative}} \cdot \underbrace{\chi(s)}_{\substack{\text{total spin} \\ \text{singlet, antisym.}}} \quad \text{Thus } \Psi(\vec{r}) \text{ should be symmetric so that } \Psi(\vec{r}_1, \vec{r}_2) \text{ is antisym.}$$

thus the electrons spend more time between the two nucleus and help to bond the molecule covalently.