

* Hamiltonian for an atom with Z protons (and electrons)

$$\mathcal{H} = \sum_{j=1}^Z \left\{ \underbrace{-\frac{\hbar^2}{2m} \nabla_j^2 - \frac{1}{4\pi\epsilon_0} \frac{Ze^2}{r_j}}_{\mathcal{H}_j \text{ (single electron)}} \right\} + \underbrace{\frac{1}{2} \frac{1}{4\pi\epsilon_0} \sum_{j \neq k}^Z \frac{e^2}{|\vec{r}_j - \vec{r}_k|}}_{\text{electron repulsion.}}$$

- will use perturbation and variational methods to deal with the electron repulsion.

- the single-particle terms are separable:

$$\text{if } \mathcal{H}_j \psi_j = E_j \psi_j \quad \text{then} \quad \mathcal{H}[\psi_1 \psi_2 \dots \psi_Z] = (E_1 + E_2 + \dots + E_Z) [\psi_1 \psi_2 \dots \psi_Z]$$

- must still antisymmetrize whole wave function.

$$\Psi(\vec{r}_1, \vec{r}_2, \dots, \vec{r}_Z, \vec{s}_1, \vec{s}_2, \dots, \vec{s}_Z) = \psi(\vec{r}_1, \vec{r}_2, \dots, \vec{r}_Z) \chi(\vec{s}_1, \vec{s}_2, \dots, \vec{s}_Z) \quad \left. \begin{array}{l} \vec{r}_i = \text{position} \\ \vec{s}_i = \text{spin} \end{array} \right\} \text{ of } i^{\text{th}} \text{ electron}$$

$$P_{ij} \Psi = P_{ij} \psi \cdot P_{ij} \chi = \pm \psi \cdot \mp \chi = -\Psi \quad \text{for fermions}$$

• thus two valid combinations: $\psi_S \chi_A$ & $\psi_A \chi_S$ (other possibilities too!)

(different states, for example consider 2 electrons.

in states ψ_a, ψ_b , either with spin \uparrow or \downarrow

$$\left. \begin{array}{l} \psi_a(\vec{r}_1) \chi_{\uparrow}(\vec{s}_1) \cdot \psi_b(\vec{r}_2) \chi_{\uparrow}(\vec{s}_2) \\ \psi_a(\vec{r}_1) \chi_{\uparrow}(\vec{s}_1) \cdot \psi_b(\vec{r}_2) \chi_{\downarrow}(\vec{s}_2) \\ \psi_a(\vec{r}_1) \chi_{\downarrow}(\vec{s}_1) \cdot \psi_b(\vec{r}_2) \chi_{\uparrow}(\vec{s}_2) \\ \psi_a(\vec{r}_1) \chi_{\downarrow}(\vec{s}_1) \cdot \psi_b(\vec{r}_2) \chi_{\downarrow}(\vec{s}_2) \end{array} \right\} \Rightarrow \begin{array}{l} \psi_A \left\{ \begin{array}{l} [\psi_a(\vec{r}_1)\psi_b(\vec{r}_2) - \psi_b(\vec{r}_1)\psi_a(\vec{r}_2)] [\chi_{\uparrow}(\vec{s}_1)\chi_{\uparrow}(\vec{s}_2)] \\ [\psi_a(\vec{r}_1)\psi_b(\vec{r}_2) - \psi_b(\vec{r}_1)\psi_a(\vec{r}_2)] [\chi_{\uparrow}(\vec{s}_1)\chi_{\downarrow}(\vec{s}_2) + \chi_{\downarrow}(\vec{s}_1)\chi_{\uparrow}(\vec{s}_2)] \\ [\psi_a(\vec{r}_1)\psi_b(\vec{r}_2) - \psi_b(\vec{r}_1)\psi_a(\vec{r}_2)] [\chi_{\downarrow}(\vec{s}_1)\chi_{\downarrow}(\vec{s}_2)] \end{array} \right\} \begin{array}{l} S=1 \\ \text{triplet} \end{array} \chi_S \\ \psi_S \left\{ [\psi_a(\vec{r}_1)\psi_b(\vec{r}_2) + \psi_b(\vec{r}_1)\psi_a(\vec{r}_2)] [\chi_{\uparrow}(\vec{s}_1)\chi_{\downarrow}(\vec{s}_2) - \chi_{\downarrow}(\vec{s}_1)\chi_{\uparrow}(\vec{s}_2)] \right\} \begin{array}{l} S=0 \\ \text{singlet} \end{array} \chi_A \end{array}$$

• note that 4 states of the form $\psi_b(\vec{r}_1) \chi(\vec{s}_1) \psi_a(\vec{r}_2) \chi(\vec{s}_2)$ disappeared due to antisymmetry.

• there is only one state with $a=b$ since $\psi_a \psi_a$ symmetric:
 $\psi_a(\vec{r}_1) \psi_a(\vec{r}_2) (\uparrow\downarrow - \downarrow\uparrow)$

* Helium atom: $\Psi(\vec{r}_1, \vec{r}_2) = \Psi_{nlm}(\vec{r}_1) \Psi_{n'l'm'}(\vec{r}_2)$ $\chi(\vec{s}_1, \vec{s}_2) = \begin{cases} |1, m\rangle & \text{triplet} \\ |0, 0\rangle & \text{singlet} \end{cases}$

• $E \propto \frac{Z}{r} \propto Z^2$ since $r \propto \frac{1}{Z}$ so $E = 4(E_n + E_{n'})$

$\Psi_0 = \Psi_{100}(\vec{r}_1) \Psi_{100}(\vec{r}_2) = \frac{8}{\pi a^3} e^{-2(r_1+r_2)/a}$ $\chi_0 = |0, 0\rangle = (\uparrow\downarrow - \downarrow\uparrow)$

Ψ_0 symmetric $\rightarrow \chi_0$ antisymmetric (singlet) : **PARAHELIUM**

$E_0 = -8 \cdot 13.6 \text{ eV} = -109 \text{ eV}$ singlet vs. -79 eV singlet observed.
(interaction energy = 30 eV)


• excited states may be either be:

$[\Psi_{nlm}(\vec{r}_1) \Psi_{100}(\vec{r}_2) + \Psi_{100}(\vec{r}_1) \Psi_{nlm}(\vec{r}_2)] |0, 0\rangle$ **PARAHELIUM**

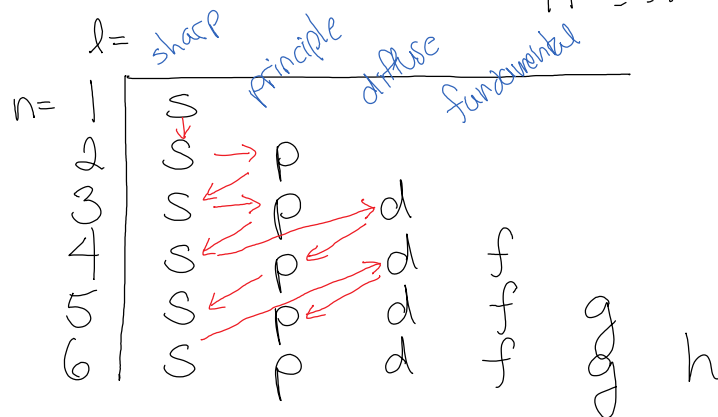
$[\Psi_{nlm}(\vec{r}_1) \Psi_{100}(\vec{r}_2) - \Psi_{100}(\vec{r}_1) \Psi_{nlm}(\vec{r}_2)] |1, \frac{1}{2}\rangle$ **ORTHOHELIUM**

para helium: 1S, 2S, 3S, 4S, ...
2P, 3P, 4P, ...
3D, 4D, ...
4F, ...

• capital letters:
TOTAL N, L, M
• same as single
excited electron
n, l, m

ortho helium:  2S, 3S, 4S, ...
2P, 3P, 4P, ...
3D, 4D, ...
4F, ...

* Atoms



spectroscopic
term

$2s+1 L_J$