

Ensemble average vs time average

$F(p, q)$: some dynamical variable
 \rightarrow no explicit time dependence

Ensemble average $\langle F \rangle(t) = \frac{1}{N} \sum_{k=1}^N F^k(t)$

$F^k(t)$ - value of this quantity for the k -th member of ensemble

When microstates are labelled by continuous parameters - e.g., (p_0, q_0) for molecules of f.a.s

$$\sum_k \rightarrow \int d^{Nd} p d^{Nd} q$$

However, technically the number of points in any volume of phase space is infinite. ∇

To get a reasonable definition need another assumption.

\Rightarrow * Divide up phase space into cells - each with a volume h_0^{Nd}

* h_0 is a (dimensional) quantity introduced by hand

* Declare: all points in this cell correspond to the same state

21.

$$\sum_k \rightarrow \int \frac{d^{Nd} p}{h_0^{Nd}} \frac{d^{Nd} q}{h_0^{Nd}}$$

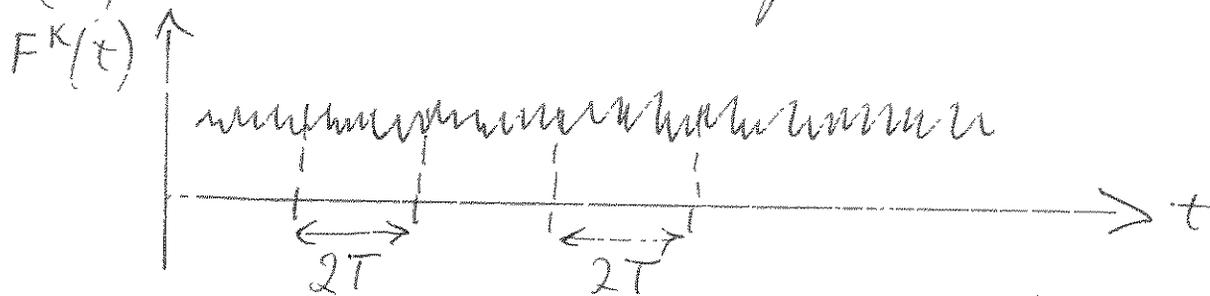
In classical theory, h_0 is some artefact used to count states

We will see that in quantum mechanics h_0 becomes Planck constant

Equilibrium $\langle F(t) \rangle$ is time independent

Consider now the behavior of a single member of the ensemble:

$F(p, q)$ clearly evolves in time
 Even in equilibrium an individual $F^k(t)$ will have time dependence.



Break up the time into large intervals of $2T$ each. The time average is then defined as

Time average: $\underbrace{\{F^k(t)\}}_{\text{ensemble average}} \equiv \frac{1}{2T} \int_{t-\epsilon}^{t+\epsilon} dt' F^k(t')$

Note, $\frac{1}{N} \sum_k \{F^k(t)\} = \frac{1}{N} \sum_k \frac{1}{2T} \int_{t-\epsilon}^{t+\epsilon} dt' F^k(t') =$
 $= \frac{1}{2T} \int_{t-\epsilon}^{t+\epsilon} dt' \left[\frac{1}{N} \sum_k F^k(t') \right] = \{ \langle F^k(t) \rangle \}$

stationary situations

In these situations a typical $F(p, q)$ has the property:

- * The behavior of $F_k(t)$ [$F_k(p, q, t)$] is similar for any sufficiently large time interval $\sim 2T$.

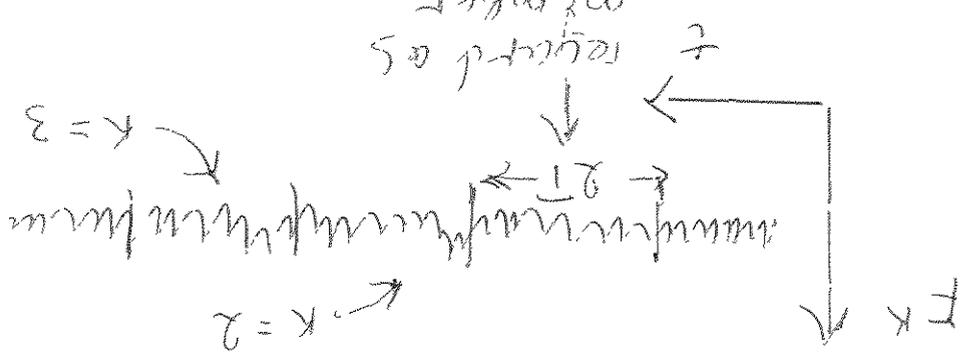
Ergodic Hypothesis

For "typical" systems, the function $F_k(t)$ for each system in the ensemble will pass through all the values accessible to it in the course of a sufficiently long time.

- * If this holds — for some given member of ensemble, and for large enough T , snapshots in time intervals

$(t_0 - T, t_0 + T)$ → should look
 $(t_0 + T, t_0 + 2T)$ → very similar

- * Similarly, each such time interval $2T$ should be representative of a member of the ensemble



Provided we choose T large enough, time average same for all members of ensemble.

(1) $\{F^k(t)\} = \{F\}$ independent of k

Similarly, ensemble average is the same for all time intervals

(2) $\langle F(t) \rangle = \langle F \rangle$ independent of time

Ensemble average of (1) $\langle \{F^k(t)\} \rangle = \{F\}$

Time average of (2)

$\langle \langle F(t) \rangle \rangle = \langle F \rangle$

$\langle F \rangle = \langle F \rangle$

\Rightarrow Time average = ensemble average

Comments:

* There are many systems which are not ergodic \rightarrow these are not integrable

* While the exact system is hard to solve \rightarrow hard to model \rightarrow only H examples

The Basic postulate → a priori probability

How we determine probabilities?

In Stat Mech the basic postulate is

All states (microstates) which can lead to a specified macrostate are equally probable

Example: 3 spins in an external H

$$\text{energy: } \epsilon = \begin{cases} +\mu H & \uparrow \\ -\mu H & \downarrow \end{cases}$$

Suppose total energy is $E = +\mu H$.
This is the only specification of the macrostate. Possible microstates & probability:

spin 1	spin 2	spin 3	probability
+	+	-	1/3
+	-	+	1/3
-	+	+	1/3

Important: probability of an individual spin is not the same for up and down state:

In fact, prob. to be + is 2/3 } for single
prob. to be - is 1/3 } spin!

Constraints on Probabilities

Microcanonical ensemble - all states of the same E are equally probable.

If system is composed of many subsystems: does this mean that each subsystem is equally likely to be in each of its possible states?

See Example with spins of 3 particles (p. 27a)

Generally, Probability for occurrence in some state } = $\frac{1}{N}$ (nu. of members of ensemble in this state)

when apart from E , some other parameter is specified, $\{y_k\}$

$\Rightarrow \Omega(E, y_k) = \left\{ \begin{array}{l} \text{subset of all states} \\ \text{of energy } E \\ \text{conditional } y_k \end{array} \right.$

$$P(y_k) = \frac{\Omega(E, y_k)}{\Omega(E)}$$

Systems with Continuous States

For systems with states specified by a set of continuous variables, e.g., particles in a large box — we need to modify the statement

Naively, in equilibrium

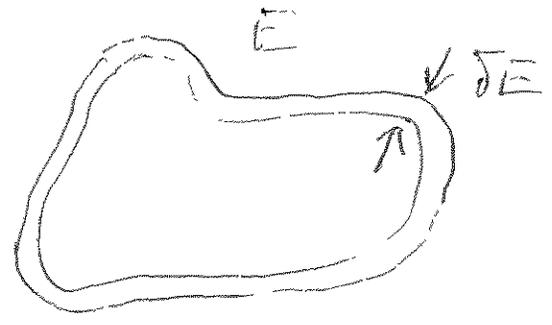
$$S(\Gamma, q) = c \delta[H(\Gamma, q) - E]$$

To make better sense of this we modify this by

$$S = \begin{cases} \text{const.} & (E - \delta E) \leq H(\Gamma, q) \leq E + \delta E \\ 0 & \text{otherwise} \end{cases}$$

$\Omega(E)$ = nu. of states with energy in the window $(E - \delta E, E + \delta E)$ =

$$= \int_{\delta E} d^N p d^N q$$



26) Microcanonical Ensemble (NR gas)

Appropriate for isolated systems

$$P(\vec{x}) = \begin{cases} C \delta[H(\vec{x}) - E] & \text{for } \vec{x} \\ 0 & \text{otherwise} \end{cases}$$

C - normalization factor

$$E - \delta E \leq H(\vec{x}) \leq E + \delta E \Rightarrow \text{probability is the same}$$

Example:

N distinguishable particles in d-space dimensions (e.g. ions in crystal)

Basic unit in phase space volume: \sqrt{h}

$$\sum_{\text{points in phase space}} \rightarrow \frac{1}{(h^{1/2})^{2Nd}} \int d^{2Nd} x$$

\Rightarrow Normalization of Ω :

$\tilde{\Omega}$ - volume not normalized

$$\Omega = \frac{\tilde{\Omega}}{(h^{1/2})^{Nd}} \text{ - normalized volume}$$

$$\text{why?} \rightarrow \Omega = \frac{1}{(h^{1/2})^{Nd}} \int d^{2Nd} x = \frac{[Pq]^{Nd}}{A} = \frac{[kg \cdot m^2]^{Nd}}{[J]^{Nd}}$$

$$\Rightarrow \left\{ \begin{array}{l} A = \frac{kg \cdot m^2}{J} \\ [h] = \frac{kg \cdot m^2}{J \cdot s} \end{array} \right\} \text{ [action] = [energy] [time] = \text{joule} \cdot \text{sec.}}$$

$$\Rightarrow A = h^{Nd} \quad \text{or} \quad (h^d)^{2Nd}$$

Because of equal probabilities



$\Omega \sim$ nu. of states

$$\Rightarrow \text{Set } \Omega(E) = \frac{1}{h^{Nd}} \int_{E \leq H(x) \leq E + \delta E} d^{2Nd} x$$

to make Ω - dimensionless $\Rightarrow x$ [action]^{-3N}

\Rightarrow counts nu. of particles in energy shell

Note: factor $1/h^{Nd}$ in front of \int , because $\Omega(E)$ must be dimensionless for entropy.

Define entropy:

$$S = k_B \ln \Omega(E)$$

when identical particles:

$$\left. \begin{array}{l} \text{particle 1 at } \vec{x}_1 \\ \text{--- " --- 2 at } \vec{x}_2 \end{array} \right\} = \left\{ \begin{array}{l} \text{particle 2 at } \vec{x}_1 \\ \text{--- " --- 1 at } \vec{x}_2 \end{array} \right.$$

$$\Rightarrow \Omega_{\text{identical}} = \frac{1}{N!} \Omega_{\text{distinguishable}}$$

$\delta E =$ Tolerance (not needed in classical treatment)

28)

E-surface: $\Gamma(E)$ = nu. of points with E

volume of phase space: $\Omega(E)$ = nu. of points in shell

For discrete spectrum, $\Omega(E)$ is not a smooth function

However, in thermo limit, when

$$\frac{E}{N} < \delta E \ll E$$

δE should be irrelevant

(In fact $\frac{\delta E}{E} = \mathcal{O}\left(\frac{1}{\sqrt{N}}\right)$)

28) Volume and area of n-dimensional sphere

(Hyper-sphere)

R-radius, Volume:

$$V_n(R) = \int_{x_1^2 + \dots + x_n^2 \leq R^2} dx_1 \dots dx_n = C_n R^n \quad (1)$$

Just from a dimensional analysis

Corresponding area: $S_{n-1}(R)$ (Focus points $x_1^2 + \dots + x_n^2 = R^2$)

We can add shells:

$$\begin{cases} V_n(R) = \int_0^R S_{n-1}(r) dr \\ S_{n-1}(R) = \frac{dV_n(R)}{dR} = n C_n R^{n-1} \end{cases} \quad (2) \quad (3)$$

Need to compute $C_n \Rightarrow$ separate (1)=(2) and use (3)

$$\int \dots \int dx_1 \dots dx_n = n C_n \int_0^R r^{n-1} dr \rightarrow \text{solve in rectangular} \quad (4)$$

or hyperspherical coordinates

In n-dimensions: $dx_1 \dots dx_n = r^{n-1} dr d\Omega_{n-1}$ (5)

For example: for $n=2 \Rightarrow d\Omega_1 = d\theta$
 $d=2 \Rightarrow d\Omega_2 = \text{side of } \theta \text{ variables}$
 \Rightarrow all ang. variables sit in constants (if integrate over n -dim. spherically-symmetric function)